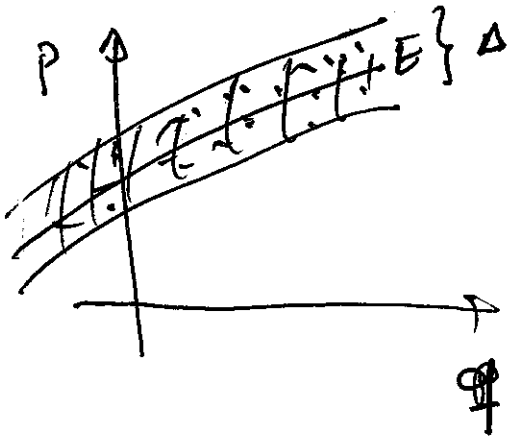


Classical statistical mechanics

(1)

Microcanonical ensemble



All states with energy in $(E - \frac{\Delta}{2}, E + \frac{\Delta}{2})$

$\Gamma(E)$: number of states.
($\Delta \ll E$)

*We can think as a set of points in phase space

The density of points stays the same along a "stream line". (Liouville theorem).

$$\rho(p, q, t) d^{3N} p d^{3N} q : \# \text{ of states in } d^{3N} p d^{3N} q$$

$$\rho(p, q) t)$$

(2)

$$\frac{dp}{dt} = \frac{\partial p}{\partial t} + \frac{\partial p}{\partial q_i} \dot{q}_i + \frac{\partial p}{\partial p_i} \dot{p}_i$$

$$\frac{\partial p}{\partial t} + \frac{\partial(\rho \dot{q}_i)}{\partial q_i} + \frac{\partial(\rho \dot{p}_i)}{\partial p_i} = 0$$

$$= \frac{dp}{dt} + \rho \frac{\partial \dot{q}_i}{\partial q_i} + \rho \frac{\partial \dot{p}_i}{\partial p_i} = \frac{dp}{dt} + \rho \frac{\partial^2 H}{\partial q_i \partial p_i} - \rho \frac{\partial^2 H}{\partial p_i \partial q_i}$$

$$\dot{q}_i = \frac{\partial H}{\partial p_i}$$

$$\dot{p}_i = - \frac{\partial H}{\partial q_i}$$

$$\frac{dp}{dt} = 0$$

"incompressible fluid"

If $\rho(q, p) = \rho(H(p, q))$

$$\Rightarrow \frac{\partial \rho}{\partial q_i} \dot{q}_i + \frac{\partial \rho}{\partial p_i} \dot{p}_i = \frac{\partial \rho}{\partial H} \left(\frac{\partial H}{\partial q_i} \dot{q}_i + \frac{\partial H}{\partial p_i} \dot{p}_i \right) - \frac{\partial \rho}{\partial H} \left(\frac{\partial H}{\partial p_i} \dot{q}_i + \frac{\partial H}{\partial q_i} \dot{p}_i \right) = 0.$$

$\Rightarrow \frac{\partial \rho}{\partial t} = 0$ "equilibrium".

If the distribution depends only on the energy \Rightarrow is indep of time.

A given system will explore a surface of constant E as a function of time.

We replace the values of local quantities by their average over the ensemble.

$$\langle f \rangle = \frac{\int d^{3N} q d^{3N} p f(p, q) \rho(p, q)}{\int d^{3N} q \int d^{3N} p \rho(p, q)}$$

If fluctuations are small :

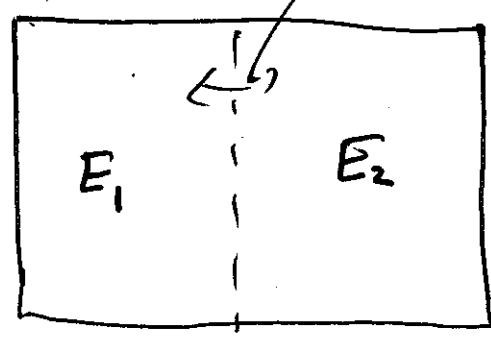
$$\frac{\langle f^2 \rangle - \langle f \rangle^2}{\langle f \rangle^2} \ll 1$$

⇒ It does not matter how we take the average ($\rho(p, q)$)

Average over all states $\rho(p, q) = 1$.

or most likely (value of f that most states have)

etc. energy can be interchanged (heat)



$$\Gamma_{12}(E_1 + E_2) = \Gamma_1(E_1) \cdot \Gamma_2(E_2)$$

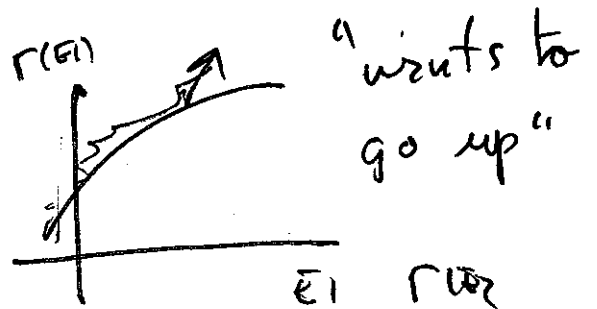
If we think $E_1 + E_2 = E$ but E_1, E_2 can change then the most likely case is the one where

$\Gamma_{12}(E_1 + E_2)$ is maximum. (like entropy!) (5)

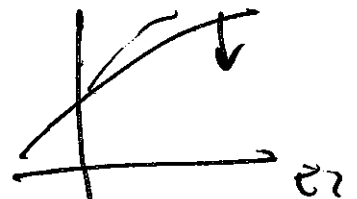
$$\Gamma_{12}(E) = \Gamma_1(E_1) \Gamma_2(E - E_1)$$

$$\frac{\partial \Gamma_{12}(E)}{\partial E_1} = \frac{\partial \Gamma_1}{\partial E_1} \Gamma_2 - \Gamma_1 \frac{\partial \Gamma_2}{\partial E_2} = 0$$

$$\frac{\partial \ln \Gamma_1}{\partial E_1} = \frac{\partial \ln \Gamma_2}{\partial E_2}$$



$$dS = \frac{1}{T} dE + \frac{P}{T} dV$$



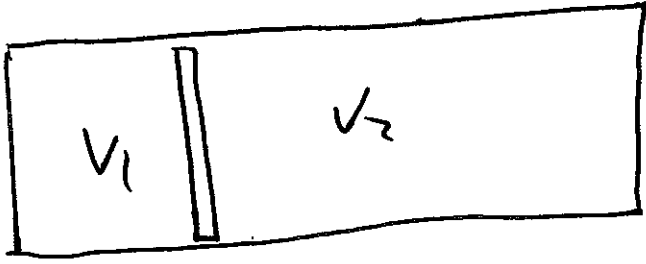
$$S = k_B \ln \Gamma(E) \quad ; \quad \frac{1}{T} = k_B \frac{\partial \ln \Gamma(E)}{\partial E}$$

$\Rightarrow T_1 = T_2$ equilibrium!

Also $\ln \Gamma_{12} = \ln \Gamma_1 + \ln \Gamma_2$
 \uparrow
 extensive!

The same happens with volume

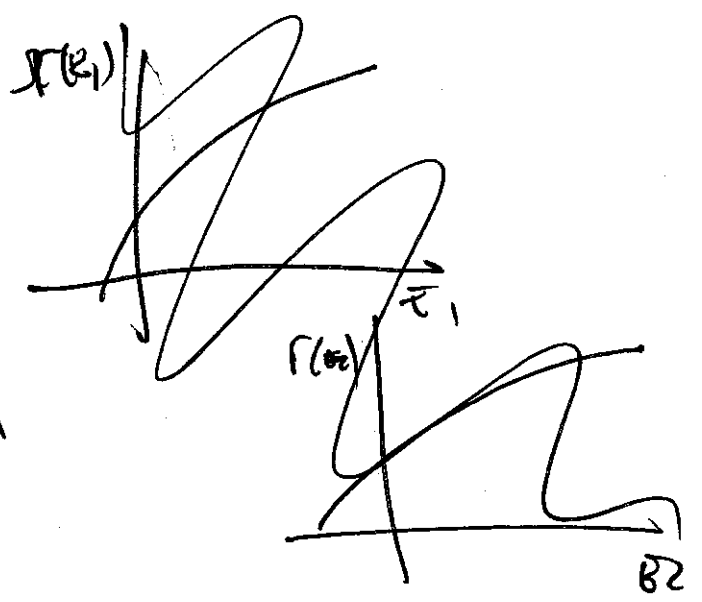
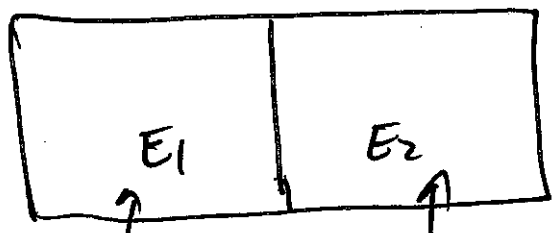
(6)



$$V_1 + V_2 = V$$

The piston moves until the # of states is maximum. (most likely scenario).

Canonical ensemble



can fluctuate now

much bigger

f_1 some property of system 1. depends on E_1

$$\langle f_1 \rangle = \frac{\int_0^{\infty} dE_1 f(E_1) \Gamma_1(E_1, E_2)}{\int_0^{\infty} \int_0^{\infty} \Gamma_1(E_1, E_2)}$$

$$= \frac{\int_0^{\infty} dE_1 f(E_1) \Gamma_1(E_1) \Gamma_2(E - E_1)}{\int_0^{\infty} \int_0^{\infty} \Gamma_1(E_1, E_2)}$$

$$\Gamma_2(E - E_1) = e^{-\frac{1}{k_B} S_2(E - E_1)} \approx e^{-\frac{S_2(E) - \frac{E}{k_B}}{k_B T_2}} = e^{-\frac{S_2(E)}{k_B T_2}} e^{\frac{E}{k_B T_2}}$$

$$\langle f_1 \rangle = \frac{\int_0^{\infty} dE_1 e^{-\frac{E_1}{k_B T_1}} \Gamma_1(E_1) f(E_1)}{\int_0^{\infty} dE_1 e^{-\frac{E_1}{k_B T_1}} \Gamma_1(E_1)}$$

$$Z = \frac{\int d^N p d^N q}{N! h^{3N}} e^{-\beta H(p, q)}$$



$$\beta = \frac{1}{k_B T}$$

partition function.

what is Z?

$$Z = \int_0^\infty dE \Gamma(E) e^{-E/k_B T}$$

$$= \int_0^\infty dE e^{\frac{1}{k_B} S(E) - E/k_B T}$$

$$= \int_0^\infty dE e^{-\frac{1}{k_B T} (E - TS)}$$

$$= \int_0^\infty dE e^{-\beta A(E)}$$

Assuming fluctuations are small smallest $A(E)$

contribution $Z \approx C e^{-\beta A}$

$$\ln Z = \ln C - \beta A$$

$$A = -\frac{1}{\beta} \ln Z + \frac{1}{\beta} \ln C$$

goes away in thermodynamic limit

$$A = -k_B T \ln Z$$

We check energy fluctuations.

$$\langle E \rangle = \frac{\int_0^\infty dE \Gamma(E) E e^{-\beta E}}{\int_0^\infty dE \Gamma(E) e^{-\beta E}} = -\frac{\partial}{\partial \beta} \ln \int_0^\infty dE \Gamma(E) e^{-\beta E}$$

$$= +\frac{\partial}{\partial \beta} (+\beta A) = A + \beta \frac{\partial A}{\partial \beta}$$

$$\frac{\partial A}{\partial \beta} = -\frac{1}{k_B \beta^2} \frac{\partial A}{\partial T}$$

$$T = \frac{1}{\beta k_B}$$

$$\langle E \rangle = A + \beta \left(-\frac{1}{k_B \beta^2} \frac{\partial A}{\partial T} \right) = A - T \frac{\partial A}{\partial T}$$

$$dA = -SdT - pdV$$

$$E = A + TS \quad \checkmark$$

Fluctuations

$$\langle (E - \langle E \rangle)^2 \rangle = \langle E^2 \rangle - \langle E \rangle^2$$

$$\langle E^2 \rangle = \frac{\int_0^\infty dE \Gamma(E) E^2 e^{-\beta E}}{\int_0^\infty dE \Gamma(E) e^{-\beta E}} = \frac{\frac{\partial^2}{\partial \beta^2} \int_0^\infty dE \Gamma(E) e^{-\beta E}}{\int_0^\infty dE \Gamma(E) e^{-\beta E}}$$

$$= \frac{\frac{\partial^2}{\partial \beta^2} e^{-\beta A}}{e^{-\beta A}} = \frac{1}{e^{-\beta A}} \frac{\partial}{\partial \beta} \left\{ e^{-\beta A} \underbrace{\left(-A - \beta \frac{\partial A}{\partial \beta} \right)}_{-E} \right\}$$

$$= \frac{1}{e^{-\beta A}} \frac{\partial}{\partial \beta} (E e^{-\beta A}) = - \frac{1}{e^{-\beta A}} \frac{\partial E}{\partial \beta} e^{-\beta A} -$$

$$- E \frac{\frac{\partial}{\partial \beta} (e^{-\beta A})}{e^{-\beta A}} = E^2 + \frac{\partial E}{\partial \beta}$$

$$\langle \dot{E}^2 \rangle - \langle E \rangle^2 = - \frac{\partial E}{\partial \beta} = \frac{1}{k_B \beta^2} \frac{\partial E}{\partial T} = k_B T^2 \frac{\partial E}{\partial T} \quad (7)$$

$$\frac{k_B}{k_B^2 T^2}$$

$$\langle (E - \langle E \rangle)^2 \rangle = k_B T^2 C_V$$

$$\sqrt{\langle (E - \langle E \rangle)^2 \rangle} = \delta E = \sqrt{k_B T^2 C_V}$$

$$\frac{\delta E}{E} \sim \frac{\sqrt{N}}{N} \sim \frac{1}{\sqrt{N}} \sim 10^{-11}$$

$N \sim 10^{23}$

Going back to how we showed $Z = e^{\beta A}$ (12)

$$Z = \int_0^{\infty} dE e^{\beta A(E)}$$

$A(E)$ minimum for \bar{E} . (β fixed)

$A(\bar{E})$

$$A(E) = A(\bar{E}) + \underbrace{\frac{\partial A}{\partial E}}_0 (E - \bar{E}) + \frac{1}{2} \frac{\partial^2 A}{\partial E^2} (E - \bar{E})^2 + \dots$$

$$A = E - TS$$

$$\frac{\partial A}{\partial E} = 1 - T \underbrace{\frac{\partial S}{\partial E}}_{1/T} = 0 \quad \checkmark$$

$$\frac{\partial^2 A}{\partial E^2} = -T \frac{\partial^2 S}{\partial E^2} = -T \frac{\partial (1/T)}{\partial E} =$$

$$= T \frac{1}{T^2} \frac{\partial T}{\partial E} = \frac{1}{C_V T}$$

$$Z = \int_0^{\infty} dE \, e^{-\beta A(\bar{E}) - \frac{1}{2C_V k_B T^2} (E - \bar{E})^2} \quad (13)$$

↑ gaussian.

$$= e^{-\beta A(\bar{E})} \sqrt{\frac{2\pi C_V k_B T^2}{k_B}}$$

$$\ln Z = -\beta A + \frac{1}{2} \ln(\pi C_V k_B T^2)$$

↑
N

ln N

ignore as $N \rightarrow \infty$

Each state has probability $P_n = \frac{e^{-\beta E_n}}{Z}$

$$\langle f \rangle = \sum_n \frac{f_n e^{-\beta E_n}}{Z}$$

↑
states.

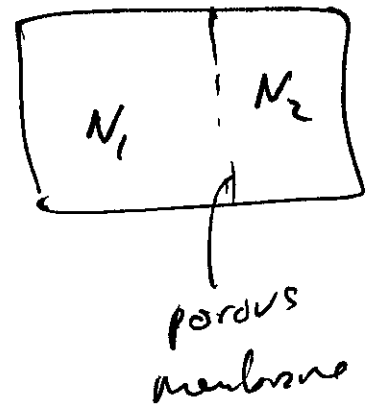
$$S \stackrel{?}{=} -k_B \sum_n P_n \ln P_n = -k_B \langle \ln P_n \rangle$$

$$= -k_B \sum_n \frac{e^{-\beta E_n}}{Z} (-\beta E_n - \ln Z) = \beta k_B \langle E \rangle + k_B \ln Z = \frac{E - A}{T} \checkmark$$

$$S = -k_B \sum_n P_n \ln P_n$$

Grand canonical ensemble.

Allows N to fluctuate



$$\Xi = \sum_{N=0}^{\infty} e^{+\beta\mu N} Z_N(V, T)$$

\uparrow
 canonical partition function.

μ : chemical potential

$$\Xi = \int_0^{\infty} dN e^{\beta\mu N - \beta A}$$

$$= \int_0^{\infty} dN e^{-\beta(A - \mu N)}$$

$$\ln \Xi = -\beta A + \beta\mu N$$

$$= -\beta(E - TS) + \beta\mu N$$

$$= \frac{1}{k_B} S - \frac{1}{k_B T} E + \frac{\mu}{k_B T} N$$

Gibbs-Duhem relation.

Suppose we have a function such that

$$f(\lambda x_1, \dots, \lambda x_n) = \lambda^n f(x_1, \dots, x_n)$$

$$\frac{\partial}{\partial \lambda} \sum_j x_j \frac{\partial f}{\partial x_j} (\lambda x_1, \dots, \lambda x_n) = n \lambda^{n-1} f(x_1, \dots, x_n)$$

$$\text{take } \lambda = 1 \Rightarrow \sum_j x_j \frac{\partial f}{\partial x_j} (x_1, \dots, x_n) = n f(x_1, \dots, x_n)$$

Consider $E(S, V, N)$ ($dE = TdS - pdV + \mu dN$)

extensive $\Rightarrow E(\lambda S, \lambda V, \lambda N) = \lambda E(S, V, N)$

$$\begin{aligned} \Rightarrow E(S, V, N) &= S \frac{\partial E}{\partial S} + V \frac{\partial E}{\partial V} + N \frac{\partial E}{\partial N} \\ &= TS - pV + \mu N \end{aligned}$$

for extensive systems.

$$\Rightarrow k_B T \ln \Sigma = TS - E + \mu N = pV$$

$$\boxed{\frac{pV}{T} = k_B \ln \Sigma}$$

Now states can have 2ng E, N

(16)

$$P_n = \frac{e^{-\beta E_n + \beta \mu N_n}}{\Xi}$$

$$\langle f \rangle = \sum_n \frac{f_n e^{-\beta E_n + \beta \mu N_n}}{\Xi}$$

$$-k_B \sum_n P_n \ln P_n = -k_B \sum_n \frac{e^{-\beta E_n + \beta \mu N_n}}{\Xi} (\beta E_n + \beta \mu N_n - \ln \Xi)$$

$$= \beta k_B E - \beta \mu k_B N + k_B \ln \Xi$$

$$= \frac{E}{T} - \frac{\mu}{T} N + \frac{PV}{T} = \frac{1}{T} (E + PV - \mu N) = \frac{TS}{T}$$

$$= S \quad \checkmark$$

also works.

Ideal gas

$$Z = \frac{\int d^{3N}p d^{3N}q}{h^{3N} N!} e^{-\beta \sum_j \frac{p_j^2}{2m}}$$

$$= \frac{V^N}{h^{3N} N!} \left(\sqrt{\frac{\pi 2m}{\beta}} \right)^{3N}$$

$$= \frac{1}{N!} \left(\frac{V}{h^3} \left(\frac{2m\pi}{\beta} \right)^{3/2} \right)^N$$

$$\ln Z = -N \ln N + N + N \ln \left(\frac{V}{h^3} \right) + \frac{3}{2} N \ln \left(\frac{2m\pi}{\beta} \right)$$

$$= N \ln \left(\frac{V}{N h^3} \right) + N + \frac{3}{2} N \ln \left(\frac{2m\pi}{\beta} \right)$$

$$\langle E \rangle = - \frac{\partial}{\partial \beta} \ln Z = \frac{3}{2} \frac{N}{\beta} = \frac{3}{2} N k_B T \quad ; \quad \boxed{C_V = \frac{3}{2} N k_B}$$

$$\beta = \frac{3}{2} \frac{N}{E}$$

$$-\beta A = N \ln \left(\frac{V}{Nh^3} \right) + N + \frac{3}{2} N \ln \left(\frac{2mT}{\beta} \right)$$

$$-\beta E + \beta TS = N \ln \left(\frac{V}{Nh^3} \right) + N + \frac{3}{2} N \ln \left(\frac{2mT}{\beta} \right)$$

$$\frac{1}{k_B} S = \frac{3}{2} N + N \ln \left(\frac{V}{Nh^3} \right) + N + \frac{3}{2} N \ln \left(\frac{2mT}{\beta} \right)$$

$$dS = \frac{1}{T} dE + \frac{P}{T} dV - \frac{\mu}{T} dN \quad \text{for fixed } E.$$

~~$$\frac{\partial S}{\partial T} \frac{1}{k_B} S = \frac{5}{2} N + N \ln \left(\frac{V}{Nh^3} \right) + \frac{3}{2} N \ln \left(\frac{2mT}{3N} \right)$$~~

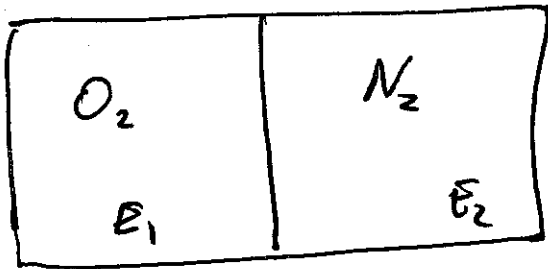
$$\left. \frac{\partial S}{\partial V} \right|_{E, N} = \frac{k_B N}{V} = \frac{P}{T} \Rightarrow \boxed{PV = Nk_B T} \checkmark$$

$$\text{or } dA = -SdT - PdV + \mu dN$$

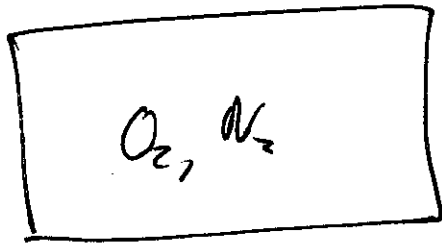
$$A = -Nk_B T \ln \left(\frac{V}{Nh^3} \right) - Nk_B T - \frac{3}{2} Nk_B T \ln \left(\frac{2mT}{\beta} \right)$$

$$P = - \left. \frac{\partial A}{\partial V} \right|_{T, N} = \frac{Nk_B T}{V} \checkmark$$

Gibbs paradox



↑ remove

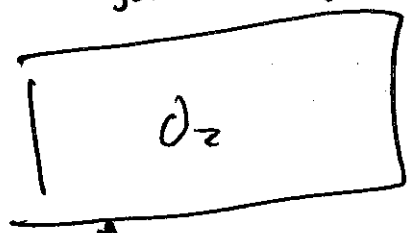
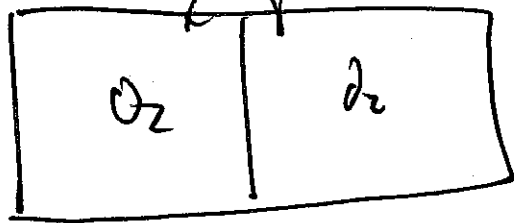


entropy increases.

each gas has more volume and it is irreversible. } ✓

Entropy of mixing

But if same density, same T



same density

entropy cannot increase on how we got

otherwise depends

Indeed

$$\frac{1}{k_B} S_1 = \frac{S}{2} N_1 + N_1 \ln \left(\frac{V_1}{N_1 h^3} \right) + \frac{3}{2} N_1 \ln \left(\frac{2\pi m_1}{\beta} \right)$$

$$\frac{1}{k_B} S_2 = \frac{S}{2} N_2 + N_2 \ln \left(\frac{V_2}{N_2 h^3} \right) + \frac{3}{2} N_2 \ln \left(\frac{2\pi m_2}{\beta} \right)$$

$$\frac{1}{k_B} S = \frac{1}{k_B} (S_1 + S_2) = \frac{S}{2} N + N \ln \left(\frac{V}{N h^3} \right) + \frac{3}{2} N \ln \left(\frac{2\pi m T}{\beta} \right)$$

↑ same density

But without $N!$

$$\frac{1}{K_B} S_1 = \frac{3}{2} N_1 + N_1 \ln \left(\frac{V_1}{h^3} \right) + \frac{3}{2} N_1 \ln \left(\frac{2\pi m_1}{\beta} \right)$$

$$\frac{1}{K_B} S_2 = \frac{3}{2} N_2 + N_2 \ln \left(\frac{V_2}{h^3} \right) + \frac{3}{2} N_2 \ln \left(\frac{2\pi m_2}{\beta} \right)$$

$$\frac{1}{K_B} S = \frac{3}{2} N + N_1 \ln \left(\frac{V_1}{h^3} \right) + N_1 \ln N_1 + \frac{3}{2} N \ln \left(\frac{2\pi m_1}{\beta} \right)$$

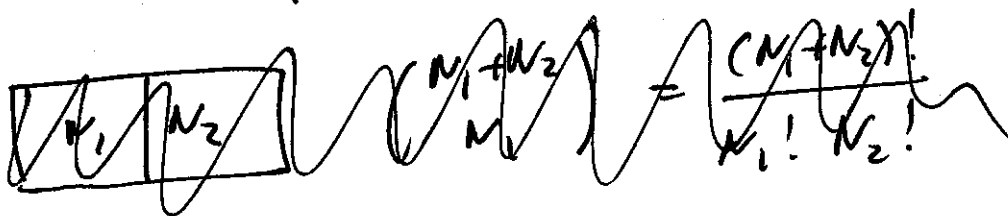
$$+ N_2 \ln \left(\frac{V_2}{h^3} \right) + N_2 \ln N_2 + \dots$$

$$= \frac{3}{2} N + N \ln \left(\frac{N}{N h^3} \right) + \frac{3}{2} N \ln \left(\frac{2\pi m_1}{\beta} \right) +$$

$$+ N_1 \ln N_1 + N_2 \ln N_2$$

instead, not extensive

Identical particles.



Equipartition theorem.

$$x_j = p_j, q_j \quad H(x_j)$$

$$\left\langle x_i \frac{\partial H}{\partial x_i} \right\rangle = \frac{1}{Z} \int \frac{d^{3N} p d^{3N} q}{N! h^{3N}} x_i \frac{\partial H}{\partial x_i} e^{-\beta H(x_j)}$$

$$= \frac{1}{Z} \frac{1}{N! h^{3N}} \int d^{3N} x_j \frac{x_i}{(-\beta)} \frac{\partial e^{-\beta H}}{\partial x_j}$$

by parts

$$= \frac{1}{Z} \frac{1}{N! h^{3N}} \int d^{3N} x_j \frac{\partial x_i}{\partial x_j} e^{-\beta H} \delta_{ij}$$

$$\left\langle x_i \frac{\partial H}{\partial x_i} \right\rangle = \delta_{ij} k_B T$$

$$\sum_i \left\langle q_i \frac{\partial H}{\partial q_i} \right\rangle = 3N k_B T$$

$\underbrace{\quad}_{-p_i}$

$$\left\langle \sum_{i=1}^{3N} q_i p_i \right\rangle = -3N k_B T$$

virial theorem

$$H = \sum_i (A_i p_i^2 + B_i q_i^2)$$

quadratic.

(23)

homogeneous. 2

A_i, B_i can
vanish.

$$\sum_i p_i \frac{\partial H}{\partial p_i} + \sum_i q_i \frac{\partial H}{\partial q_i} = 2H$$

f constants non-vanishing

$$\left\langle \underbrace{p_i \frac{\partial H}{\partial p_i}}_{\text{no sum}} \right\rangle = k_B T$$

$$\langle q_i \frac{\partial H}{\partial q_i} \rangle = k_B T.$$

$$2 \langle H \rangle = f k_B T$$

$$\langle H \rangle = \frac{1}{2} f k_B T$$

← equipartition

$\frac{1}{2} k_B T$ for each quadratic term in H.

$$C_V = \frac{1}{2} k_B f.$$

But also, for each non-vanishing A, or B

$$\frac{1}{2} \int dq B q^2 e^{-\beta B q^2} = -\frac{\partial}{\partial \beta} \ln \int dq e^{-\beta B q^2}$$

$$= -\frac{\partial}{\partial \beta} \ln \sqrt{\frac{\pi}{\beta B}} = \frac{1}{2} \frac{\partial \ln \beta}{\partial \beta} = \frac{k_B T}{2} \leftarrow \frac{1}{2} \text{ 2 degrees}$$