

### Question 1 (Kevin Brown)

1.) Minimize  $H = \int d^3p f(p) \ln(f(p))$  given  $N = \int d^3p f(p)$ ;  $E = \int d^3p \frac{p^2}{2m} f(p)$

• Lagrange multipliers:

$$\nabla_{\alpha, \beta} \left[ H - \alpha \left( \int d^3p f(p) \right) - \beta \left( \frac{1}{N} \int d^3p \frac{p^2}{2m} f(p) \right) \right] = 0$$

$$\nabla_{f, \alpha, \beta} \left[ \int d^3p \left[ f(p) \ln f(p) - \alpha f(p) - \frac{\beta}{N} \frac{p^2}{2m} f(p) \right] \right] = 0$$

$$\nabla_{\alpha, \beta} \left[ \int d^3p \left[ \ln f(p) + 1 - \alpha - \frac{\beta}{N} \frac{p^2}{2m} \right] \right] = 0$$

• Can be satisfied by letting integrand be = 0:

$$\ln f(p) + 1 - \alpha - \frac{\beta}{N} \frac{p^2}{2m} = 0 \rightarrow f(p) = e^{\alpha + \frac{\beta}{N} \frac{p^2}{2m} - 1}$$

$$\text{Let } C = e^{\alpha - 1} \rightarrow f(p) = C e^{\frac{\beta}{N} \frac{p^2}{2m}}$$

• Now use  $N = \int d^3p C e^{\frac{\beta}{N} \frac{p^2}{2m}} = 2\sqrt{2} \left( \frac{-\pi m N}{\beta} \right)^{3/2} C \rightarrow$

$$C = \frac{N}{2\sqrt{2} \pi m N} \left( \frac{-\beta}{2\pi m N} \right)^{3/2} = \left( \frac{-\beta N^{3/2}}{2\pi m N^{3/2}} \right)^{3/2} = \left( \frac{-\beta}{2\pi m N^{3/2}} \right)^{3/2}$$

• Now use  $E = \frac{1}{N} \int d^3p \frac{p^2}{2m} \left[ C e^{\frac{\beta}{N} \frac{p^2}{2m}} \right] = \frac{C}{N} \int d^3p \frac{p^2}{2m} e^{\frac{\beta}{N} \frac{p^2}{2m}} \rightarrow \frac{C}{2mN} \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} dp_x dp_y dp_z (p_x^2 + p_y^2 + p_z^2) e^{\frac{\beta}{2mN} (p_x^2 + p_y^2 + p_z^2)} = \frac{3\sqrt{2} C \pi^{3/2} \left( \frac{-mN}{\beta} \right)^{5/2}}{mN}$

$$E = \frac{3\sqrt{2} \pi^{3/2}}{mN} \left( \frac{-mN}{\beta} \right)^{5/2} \cdot \frac{N}{2\sqrt{2} \pi m N} \left( \frac{-\beta}{2\pi m N^{3/2}} \right)^{3/2} = \frac{3mN}{2m\beta} \left( \frac{-mN}{\beta} \right)^{3/2} \left( \frac{-\beta}{mN} \right)^{3/2} \rightarrow E = \frac{3N}{2\beta} \rightarrow \beta = \frac{3N}{2E} \rightarrow C = \frac{N}{2\sqrt{2} \pi m N} \cdot \left( \frac{-3N}{2E} \right)^{3/2} \rightarrow C = N \left( \frac{3}{4\pi m E} \right)^{3/2}$$

• By equipartition,  $E = \frac{3}{2} k_B T \rightarrow f(p) = \frac{N}{(2\pi m k_B T)^{3/2}} e^{-\frac{p^2}{2m k_B T}}$

• This shows that the  $f(p)$  that minimizes the H-factor is the Maxwell-Boltzmann distribution. For a classical gas in equilibrium, the distribution function will then almost always be Maxwell-Boltzmann.

Question 2 (Ankit Kundu)

PROBLEM-2

$$\text{Probability of being of devoid} = e^{-N(V/V)}$$

where  $N \equiv$  no. of particles,

$V \equiv$  volume devoid

$V \equiv$  Total volume.

Considering ideal gas,  $PV = nRT$

at  $P = 10^5 \text{ Pa}$ ,  $T = 300 \text{ K}$ ,  $V = 10 \text{ m}^3$

$$n = \frac{10^5 \times 10}{R \times 300} = \frac{10^6}{8.314459 \times 300} \text{ moles}$$

$$= 4 \times 10^3 \text{ moles}$$

$$N = n \times 6.022 \times 10^{23} \text{ molecules/mole}$$

$$= 24.088 \times 10^{26} \text{ molecules.}$$

Part 1  $V \equiv$  Volume devoid  $= 1 \text{ cm}^3 = (0.01)^3 \text{ m}^3$   
 $= 10^{-6} \text{ m}^3$

$\therefore$  Prob. of  $1 \text{ cm}^3$  being devoid

$$= e^{-24.088 \times 10^{26} \times 10^{-6}/10}$$

$= 0$  no chance almost.

Part 2  $V \equiv$  Volume devoid =  $1 \text{ \AA}^3 = (10^{-10})^3 \text{ m}^3$   
 $= 10^{-30} \text{ m}^3$

$\therefore$  Probability of  $1 \text{ \AA}^3$  being devoid

$$= e^{-24.088 \times 10^{26} \times 10^{-30} \text{ m}^3 / 10^{-30} \text{ m}^3}$$

$$= e^{-24.088 \times 10^{-3}}$$

$$= 0.976 \quad \text{high chance of being void.}$$

(Ans)

Question 3 (Kevin Brown)

$$3.) \text{ Density fluctuations: } \langle N^2 \rangle - \langle N \rangle^2 = z \frac{\partial}{\partial z} z \frac{\partial}{\partial z} \ln \Xi = kTV \frac{\partial^2 P}{\partial \mu^2}$$

$$\text{where } z = e^{\beta \mu} = \text{fugacity}, \quad \Xi = \sum_{n=0}^{\infty} z^n \sum_n e^{-\beta E_n + \beta \mu n}$$

• Helmholtz energy is extensive:  $A(N, V, T) = N a(v)$ ,  $v \equiv V/N$

$$\mu = \left[ \frac{\partial A(N, V, T)}{\partial N} \right]_{V, T} ; \quad P = - \left[ \frac{\partial A(N, V, T)}{\partial V} \right]_{N, T}$$

$$\hookrightarrow \frac{\partial \mu}{\partial v} = -v \frac{\partial^2 a(v)}{\partial v^2} \quad \frac{\partial P}{\partial \mu} = \frac{\partial P / \partial v}{\partial \mu / \partial v} = \frac{1}{v}$$

$$\frac{\partial^2 P}{\partial \mu^2} = -\frac{1}{v^2} \frac{\partial v}{\partial \mu} = \frac{1}{v^3 \partial^2 a / \partial v^2} = -\frac{1}{v^3 \partial P / \partial v} \rightarrow$$

$$\boxed{\langle N^2 \rangle - \langle N \rangle^2 = \bar{N} kT k_T / v ; \quad k_T = \frac{1}{v(-\partial P / \partial v)}}$$

• As  $\bar{N} \rightarrow \infty$ , almost all systems in the ensemble have density  $\langle N \rangle$

Question 4 (Mingi Kim)

Problem 4. Grand partition fn of ideal gas

$$\begin{aligned}
 \Xi(\mu, V, T) &= \sum_{N=0}^{\infty} z^N \left( \int \frac{d^3p}{h^3} e^{-\beta p^2} \right)^N \quad z = e^{\beta \mu} \\
 &= \sum_{N=0}^{\infty} z^N \frac{V}{h^3 N!} \left( \int d^3p e^{-\frac{\beta p^2}{2m}} \right)^N \\
 &\quad \rightarrow \left( \int d^3p e^{-\frac{\beta p^2}{2m}} \right)^3 = \left( \frac{\pi}{\beta m} \right)^{3/2} \\
 &= \sum_{N=0}^{\infty} z^N \frac{V^N}{h^{3N} N!} \left( \frac{2\pi m}{\beta} \right)^{3N/2} \\
 &= \sum_{N=0}^{\infty} \frac{V^N}{N!} z^N \left( \frac{2\pi m}{\beta h^2} \right)^{3N/2} \quad \lambda_T = \sqrt{\frac{2\pi \hbar^2}{m k_B T}} : \text{thermal wavelength} \\
 &= \sum_{N=0}^{\infty} \frac{V^N}{N!} z^N \left( \frac{m k_B T}{2\pi \hbar^2} \right)^{3N/2} \\
 &= \sum_{N=0}^{\infty} \frac{V^N}{N!} z^N \left( \frac{1}{\lambda_T^3} \right)^{3N} = \sum_{N=0}^{\infty} \frac{1}{N!} \left( \frac{V z}{\lambda_T^3} \right)^N \\
 &= \exp \left( e^{\beta \mu} \frac{V}{\lambda_T^3} \right)
 \end{aligned}$$

$$\begin{aligned}
 \Phi &= U - TS - \mu N \\
 &= -k_B T \ln \Xi = -P(\mu, T) V
 \end{aligned}$$

$$d\Phi = -SdT - PdV - Nd\mu$$

# of particles

$$\begin{aligned}
 N &= -\frac{\partial \Phi}{\partial \mu} = \frac{\partial}{\partial \mu} (-k_B T \ln \Xi) = k_B T \frac{\partial}{\partial \mu} \left( V \frac{e^{\beta \mu}}{\lambda_T^3} \right) \\
 &= \frac{V e^{\beta \mu}}{\lambda_T^3}
 \end{aligned}$$

chemical potential

$$\begin{aligned}
 \mu &= \frac{1}{\beta} \ln \left( \frac{N \lambda_T^3}{V} \right) \\
 &= k_B T \ln \left( \frac{N \lambda_T^3}{V} \right)
 \end{aligned}$$

many

$$S = - \frac{\partial \Phi}{\partial T} = \frac{\partial}{\partial T} (k_B T \ln \Xi)$$

$$= \frac{\partial}{\partial T} (k_B T e^{\mu \frac{V}{\lambda^3}})$$

$$= k_B e^{\mu \frac{V}{\lambda^3}} + k_B T e^{\mu \frac{V}{\lambda^3}} \mu \frac{\partial}{\partial T} + k_B T e^{\mu \frac{V}{\lambda^3}} \frac{\partial V}{\lambda^3} \frac{\partial \lambda}{\partial T}$$

$$= k_B e^{\mu \frac{V}{\lambda^3}} + k_B T e^{\mu \frac{V}{\lambda^3}} \mu \left(-\frac{1}{k_B T^2}\right) + k_B T e^{\mu \frac{V}{\lambda^3}} \frac{3V}{\lambda^4} \cdot \left(-\frac{\partial \lambda}{2T}\right)$$

$$= e^{\mu \frac{V}{\lambda^3}} \left( \frac{k_B V}{\lambda^3} - \frac{k_B T V}{\lambda^3} \frac{\mu}{k_B T^2} + \frac{3V}{\lambda^4} k_B T \frac{\lambda}{2T} \right)$$

$$= e^{\mu \frac{V}{\lambda^3}} \left( \frac{5}{2} \frac{k_B V}{\lambda^3} - \frac{\mu}{T \lambda^3} \right)$$

$$= \frac{e^{\mu \frac{V}{\lambda^3}}}{\lambda^3} V \left( \frac{5}{2} k_B - \frac{\mu}{T} \right)$$

$$= \left[ \frac{e^{\mu \frac{V}{\lambda^3}}}{\lambda^3} \frac{V}{T} \left( \frac{5}{2} k_B T - \mu \right) \right] = N \left( \frac{5}{2} k_B - \frac{\mu}{T} \right)$$

$$k = \frac{1}{k_B T}$$

$$\frac{\partial k}{\partial T} = -\frac{1}{k_B T^2}$$

$$\frac{\partial \lambda}{\partial T} = \sqrt{\frac{2\pi h^2}{m k_B T}} \left(-\frac{1}{2} \frac{1}{T^2}\right)$$

$$= \lambda T \times \frac{1}{2T} = -\frac{\lambda}{2T}$$

equation of state

$$\Phi = -P(\mu, T) \cdot V = -k_B T \ln \Xi = -k_B T e^{\mu \frac{V}{\lambda^3}} = -N k_B T$$

$$\therefore PV = N k_B T$$

many

$$\langle E \rangle = - \frac{\partial}{\partial \beta} \ln \Xi = - \frac{\partial}{\partial \beta} \left( e^{\mu \frac{V}{\lambda^3}} \right)$$

$$\lambda = \sqrt{\frac{2\pi h^2}{m \beta}}$$

$$\frac{\partial \lambda}{\partial \beta} = \sqrt{\frac{2\pi h^2}{m}} \frac{1}{2\beta^{3/2}} = -\frac{\lambda}{2\beta}$$

$$= - \left( \mu e^{\mu \frac{V}{\lambda^3}} \frac{V}{\lambda^3} - e^{\mu \frac{V}{\lambda^3}} \frac{3V}{\lambda^4} \frac{\partial \lambda}{\partial \beta} \right)$$

$$= - e^{\mu \frac{V}{\lambda^3}} \left( \mu \frac{V}{\lambda^3} - \frac{3V}{\lambda^4} \frac{\lambda}{2\beta} \right) =$$

$$\frac{e^{\mu \frac{V}{\lambda^3}}}{\lambda^3} V \left( \frac{3}{2} k_B T - \mu \right)$$

specific heat

$$C_V = \left. \frac{\partial \langle E \rangle}{\partial T} \right|_{V, N}$$

$$= \frac{\partial}{\partial T} \left( N \left( \frac{3}{2} k_B T - \mu \right) \right)_{V, N} = \frac{3}{2} k_B$$